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## LIGAND DISPLACEMENT REACTIONS OF $MF_5$ .L (M = P, As, Sb) STUDIED BY MULTINUCLEAR NMR SPECTROSCOPY

Michael F. A. Dove, Eleanor M. Lloyd Jones\* and Jeremy C. P. Sanders Department of Chemistry, The University, Nottingham NG7 2RD (U.K.)

The pentafluorides of phosphorus, arsenic and antimony are known to form neutral 1:1 complexes with many organic bases (E.L. Muetterties, Advances in the Chemistry of Coordination Compounds, 1961, 513). Such compounds are of interest as possible precursors in the preparation of the hexahalo anions PF<sub>5</sub>X (P.J. Chevrier and S. Brownstein, J.Inorg.Nucl.Chem., 1980, <u>42</u>, 1397), AsF<sub>5</sub>X<sup>-</sup> and SbF<sub>5</sub>X<sup>-</sup> (M.F.A. Dove, J.C.P. Sanders, E. Lloyd Jones and M.J. Parkin, J.Chem.Soc., Chem.Commun., 1984, 1578). The adducts MF<sub>r</sub>.L (M = P, As, Sb; L = MeCN, py,  $Me_3N$ ) have, therefore, been synthesised in a pure state by direct reaction of the base with the metal pentafluoride in a suitable solvent (such as MeCN,  $CH_2Cl_2$  or  $CClF_2CCl_2F$ ). The adducts were characterised by  ${}^{31}P$ ,  ${}^{75}As$ ,  ${}^{121}Sb$  and  ${}^{19}F$  NMR spectroscopy. With the exception of  ${\tt PF}_{\sf S}.{\tt MeCN}$ , the central nucleus resonances have not previously been studied. Although no <sup>75</sup>As  $(I = {}^{3}/{2})$  or  ${}^{121}$ Sb  $(I = {}^{5}/{2})$  resonances could be obtained from the acetonitrile adducts, the central nucleus resonances of the pyridine and trimethylamine adducts showed binomial sextets. Proton decoupling resolved the one bond coupling to  ${}^{14}N$  (I = 1) in PF<sub>5</sub>.NMe<sub>3</sub>, PF<sub>5</sub>.py and AsF<sub>5</sub>.py.

There is a large contrast between the reactivities of these adducts, which is shown by attempts to displace the base to form MF<sub>5</sub>x<sup>-</sup> anions. Addition of methanol to PF<sub>5</sub>.L and SbF<sub>5</sub>.L resulted in the formation of the pentafluoromethoxy anions. NMe<sub>3</sub>H<sup>+</sup>PF<sub>5</sub>OMe<sup>-</sup> and pyH<sup>+</sup>SbF<sub>5</sub>OMe<sup>-</sup> were isolated as pure solids and the anions were characterised by NMR, but no <sup>121</sup>Sb resonance was obtained from the SbF<sub>5</sub>OMe<sup>-</sup> anion. SbF<sub>5</sub>OMe<sup>-</sup> was converted to SbF<sub>5</sub>X<sup>-</sup> (X = Cl, Br) by reaction with the corresponding acetyl halide. SbF<sub>5</sub>.MeCN reacted with NEt<sub>4</sub><sup>+</sup>X<sup>-</sup> (X = Cl, Br, NCS, NCO) to form SbF<sub>5</sub>X<sup>-</sup>. The previously unobserved SbF<sub>5</sub>NCS<sup>-</sup> and SbF<sub>5</sub>NCO<sup>-</sup> anions were shown by <sup>14</sup>N NMR to be N-bonded. However, addition of NEt<sub>4</sub><sup>+</sup>X<sup>-</sup> to PF<sub>5</sub>.MeCN or AsF<sub>5</sub>.MeCN produced a mixture of MF<sub>6</sub><sup>-</sup> and MF<sub>5</sub>X<sup>-</sup>. Pyridine and trimethylamine were found to be more difficult to displace. Complete conversion to MF<sub>5</sub>X<sup>-</sup> was observed only with the antimony adducts. AsF<sub>5</sub>.py and AsF<sub>5</sub>.NMe<sub>3</sub> showed anomalous behaviour since they were air-stable, and inert to substitution of the base.